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metal spheres that have different

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amounts of. charge on them. The spheres in diagram C are closer than the others because they have the strongest. attraction.

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your responsibility to come to office hours or ask during class work time.

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Thermodynamics

HW Keys - Roosevelt High School AP
Chemistry 2017-18

This video is designed to help you
complete the molarity chemquest
found in Unit 11.

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Molarity

118 ChemQuest 39 Name: _____

Date: _____ Hour: _____ Information:

Molarity Concentration is a term that describes the amount of solute that is dissolved in a solution. Concentrated solutions contain a lot of dissolved solute, but dilute solutions contain only

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a little.

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An element exists as 4 different isotopes. 4.35% have a mass of 49.9461, 83.79% have a mass of 51.9405, 9.50% have a mass of 52.9407 amu, and 2.36% have a mass of 53.9389 amu. a. What is the average atomic mass for this element?

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b. What is the identity of this element?

Chemistry: ChemQuest Skill Practice
(Semester 1 ...

traditionally and ... Chemquest 23

Answer Key More Hint: add your

answer to question 1 and four times

your answer to question 2. $4 + 4(7) = 4$

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+ 28 = 32 4. Chemistry Inquiry
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of the options to accompany you in
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question 1 and four times your answer
to question 2. $4 + 4(7) = 4 + 28 = 32$ 4.
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15. Given your answer to question 14 and the reaction you wrote in question 13, calculate the molarity of the nitrate

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ions (NO₃⁻) in the solution. Title:
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(Base the answer to this question on
the number of moles of propane that

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actually get combusted (which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = $4(3) = 12$.

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